



University of Pennsylvania
ScholarlyCommons

Department of Physics Papers

Department of Physics


2018

The Role of Quantum Decoherence in FRET

Philip C. Nelson

University of Pennsylvania, nelson@physics.upenn.edu

Follow this and additional works at: https://repository.upenn.edu/physics_papers

 Part of the [Atomic, Molecular and Optical Physics Commons](#), [Biological and Chemical Physics Commons](#), and the [Optics Commons](#)

Recommended Citation

Nelson, P. C. (2018). The Role of Quantum Decoherence in FRET. *Biophysical Journal*, 115 167-172.
<http://dx.doi.org/10.1016/j.bpj.2018.01.010>

This paper is posted at ScholarlyCommons. https://repository.upenn.edu/physics_papers/650
For more information, please contact repository@pobox.upenn.edu.

The Role of Quantum Decoherence in FRET

Abstract

Resonance energy transfer has become an indispensable experimental tool for single-molecule and single-cell biophysics. Its physical underpinnings, however, are subtle: It involves a discrete jump of excitation from one molecule to another, and so we regard it as a strongly quantum-mechanical process. And yet, its kinetics differ from what many of us were taught about two-state quantum systems; quantum superpositions of the states do not seem to arise; and so on. Although J. R. Oppenheimer and T. Förster navigated these subtleties successfully, it remains hard to find an elementary derivation in modern language. The key step involves acknowledging quantum decoherence. Appreciating that aspect can be helpful when we attempt to extend our understanding to situations where Förster's original analysis is not applicable.

Keywords

Forster resonance transfer, resonance energy transfer

Disciplines

Atomic, Molecular and Optical Physics | Biological and Chemical Physics | Optics | Physical Sciences and Mathematics | Physics

The Role of Quantum Decoherence in FRET

Philip C. Nelson
Department of Physics and Astronomy
University of Pennsylvania
Philadelphia PA USA

Running head: Role of Quantum Decoherence in FRET

Abstract

Resonance energy transfer has become an indispensable experimental tool for single-molecule and single-cell biophysics. Its physical underpinnings, however, are subtle: It involves a discrete jump of excitation from one molecule to another, and so we regard it as a strongly quantum-mechanical process. And yet, its kinetics differ from what many of us were taught about two-state quantum systems; quantum superpositions of the states do not seem to arise; and so on. Although J. R. Oppenheimer and T. Förster navigated these subtleties successfully, it remains hard to find an elementary derivation in modern language. The key step involves acknowledging quantum decoherence. Appreciating that aspect can be helpful when we attempt to extend our understanding to situations where Förster’s original analysis is not applicable.

The phenomenon of nonradiative resonance energy transfer, often called “Förster” or “fluorescence” resonance energy transfer (FRET), has become a central tool in biophysical instrumentation [1], starting from the demonstration that it could be observed at the single-molecule level in room-temperature, aqueous solution [2]. Today FRET-based sensors detect intra- and intermolecular motions, local chemical environment, and even mechanical forces [3], both in vitro and even inside living cells, with high time resolution.

FRET displays both classical and quantum aspects. When we discuss it, we often imagine an excitation state of a fluorophore as a discrete *thing* that can be transferred from a donor to an acceptor fluorophore intact, as basketball players pass the ball. That picture never mentions the possibility of quantum-mechanical superposition states, in which an excitation is delocalized (simultaneously located on two different fluorophores), even though we know that two-state quantum systems generally do show that phenomenon. Also, most descriptions tacitly assume that the transfer can be described by a fixed probability per unit time; that is, the transfer follows a rate equation with first-order kinetics. Again, however, two-state quantum mechanical transitions generally do not behave in this classical way (see Sect. 1 below).

In short, although we might have expected behavior intermediate between the classical and quantum regimes, instead FRET seems to involve fluorophores simultaneously displaying *strongly* quantum behavior (discrete energy levels), but also *strongly* classical behavior (no superpositions, localized excitations, first-order rate equations). It is worthwhile to see how this is possible.

The original quantum derivations by Oppenheimer and Förster [4–6] appealed to the “golden rule” of quantum mechanics [8]. However, Physics textbooks present that result in a way that seems applicable only for transitions to or from a continuum of states—not the discrete electron energy levels of donor and acceptor fluorophores. Despite that discrepancy, these early derivations gave accurate results because the “golden rule” approach implicitly incorporates the crucial feature of fast quantum decoherence [9, 10]. The aim of the present note is to keep that feature in plain view, using terminology borrowed from quantum electronics, e.g. [11]. Although known to specialists, this approach is simple enough to deserve wider currency.

For historical details about the discovery of FRET, see [9, 12, 13]. More details about this derivation are given in [14].

1 Isolated two-state system

To focus on the key issues, we will not discuss the process that excites the donor, nor the eventual fluorescence of the acceptor, instead concentrating on the transfer of the excitation from one to the other. We will also make some simplifying assumptions:

- We suppose that only two electronic states of the donor are relevant: the ground state $|D_0\rangle$ and one excited state $|D_\star\rangle$. Similarly, we consider only two acceptor states $|A_0\rangle$ and $|A_\star\rangle$. We are particularly interested in transitions between joint states of the form

$$|1\rangle = |D_\star A_0\rangle, \quad |2\rangle = |D_0 A_\star\rangle, \quad (1)$$

whose energies are nearly equal (the resonance condition). Direct transitions between those two states, without any photon emission, are therefore compatible with energy conservation.

- We will eventually define a “decoherence time” T , and assume that it is much shorter than the hopping time ($T \ll \Omega^{-1}$ below). We will also assume that T is much shorter than the mean waiting time before loss processes other than FRET deexcite the donor ($T \ll \tau$ below).

The transition between the states in Eq. 1 would be easy to describe in a world containing only two atoms [15, chaps. 7–9]. Suppose for a moment that the two states of interest have exactly the same electronic-state energy in isolation (they are exactly resonant). Make the convenient convention that these energy values are $E_1 = E_2 = 0$. When the two atoms are brought near each other, they will have a coupling giving rise to a Hamiltonian operator with an off-diagonal entry in the $1, 2$ basis, which we may take to be real:

$$H = \begin{bmatrix} 0 & V \\ V & 0 \end{bmatrix}. \quad (2)$$

The system’s evolving state can then be expanded as

$$|\Psi(t)\rangle = a(t)|1\rangle + b(t)|2\rangle, \quad (3)$$

where the coefficient functions obey the Schrödinger equation:

$$i\hbar \begin{bmatrix} da/dt \\ db/dt \end{bmatrix} = V \begin{bmatrix} b \\ a \end{bmatrix}.$$

Consider the solution with the initial state $|\Psi(0)\rangle = |1\rangle$; at later times, we find that $|b(t)|^2 = \sin^2(\Omega t/2)$, where $\Omega = 2V/\hbar$. Interpreting this quantity as

the probability to find the system in state 2, we conclude that the probability initially increases with time as t^2 . But this means that the *initial* growth rate of the probability is zero, contrary to the first-order kinetics observed in FRET. Moreover, at almost every time the state is a quantum superposition of $|1\rangle$ and $|2\rangle$, in contrast to the “basketball” picture of resonance energy transfer alluded to above. Finally, the solution just found is oscillatory: The system periodically reverts to being completely in state 1, in contrast to the one-way transfer characteristic of FRET.

2 Two-state system with environment

2.1

To see where we have gone astray, we must remember that our two fluorophores are hardly alone: They are just a *subsystem* of the entire world. Each constantly suffers collisions with surrounding water molecules, as well as less obvious influences involving fluctuating electric fields in its neighborhood and so on. A good fluorophore is robust to such disturbances, in the sense that they rarely knock it into a different electronic state. Nevertheless, environmental influences can affect the quantum-mechanical *phase* of a fluorophore’s state, by momentarily perturbing its energy levels during each collision. One way to incorporate this effect is by coupling both the donor and acceptor to a “bath,” for instance, of harmonic oscillators [16,17]. But the exact nature of the bath turns out not to be very important, leading us to suspect that there must be an approach that is not so explicit, and hence is computationally simpler.

A state measurement that could in principle be made internally to the donor–acceptor subsystem \mathfrak{s} corresponds to an observable \mathbf{O} that acts only on the subsystem’s two-dimensional state space $\mathcal{H}_{\mathfrak{s}}$. Given a pure state, we can express the measured value of such an observable without needing to know anything about the environment \mathfrak{e} :

$$\langle \mathbf{O} \rangle = {}_{\mathfrak{s}}\langle \psi | \mathbf{O} | \psi \rangle_{\mathfrak{s}} \quad \text{for a pure state } |\Psi\rangle = |\psi\rangle_{\mathfrak{s}} \otimes |\phi\rangle_{\mathfrak{e}}. \quad (4)$$

Unfortunately, even if we could prepare a pure initial state, it would quickly evolve into an entangled state due to the interactions between \mathfrak{s} and \mathfrak{e} . However, we can still compactly summarize the effect of the environment on the measured values of observables that, like \mathbf{O} , refer only to the subsystem. To do this, we introduce a Hermitian operator ρ on $\mathcal{H}_{\mathfrak{s}}$ called the density operator, defined by constructing the dyad $|\Psi\rangle\langle\Psi|$ and taking the trace over the environment state space [18, 19]:

$$\rho = \text{Tr}_{\mathfrak{e}} \left(|\Psi\rangle\langle\Psi| \right). \quad (5)$$

In our problem, ρ can be represented by a two-dimensional matrix with respect to the basis $|1\rangle, |2\rangle$. If we know ρ , then the measured value of any subsystem observable can be expressed as

$$\langle \mathbf{O} \rangle = \text{Tr}_{\mathfrak{s}} (\rho \mathbf{O}) \quad \text{for any state, represented by } \rho. \quad (6)$$

In order for this formulation to be useful, we need to be able to compute ρ , at least approximately. This is not difficult when \mathfrak{s} is perfectly isolated from its environment, because in that case a pure (unentangled) state remains pure:

$$|\Psi(t)\rangle = |\psi(t)\rangle_{\mathfrak{s}} \otimes |\phi(t)\rangle_{\mathfrak{e}} \quad \text{for isolated subsystem.} \quad (7)$$

Here $|\psi(t)\rangle_{\mathfrak{s}}$ denotes the time development of the subsystem under its Hamiltonian, independent of that of the environment, $|\phi(t)\rangle_{\mathfrak{e}}$.

Still restricting to the case of an isolated subsystem, the time development of ρ is determined by $H_{\mathfrak{s}}$, the subsystem's Hamiltonian operator:

$$\frac{d\rho}{dt} = \frac{1}{i\hbar}[H_{\mathfrak{s}}, \rho] \quad \text{for isolated subsystem.} \quad (8)$$

Notice that Eqs. 3, 5, and 7 give

$$\rho(t) = |\psi(t)\rangle_{\mathfrak{s}\mathfrak{s}} \langle\psi(t)| \quad \text{so} \quad \rho_{ij} = \begin{bmatrix} |a(t)|^2 & a(t)b(t)^* \\ a(t)^*b(t) & |b(t)|^2 \end{bmatrix}_{ij}. \quad (9)$$

This formula shows that the diagonal elements of ρ (“populations”) reflect the respective probabilities to be in the two states. Unlike the off-diagonal elements (“coherences”), the populations are unaffected if we change basis states to new versions differing by phases from the old ones, for example, $|1'\rangle = e^{i\theta}|1\rangle$.

2.2

As mentioned earlier, interactions with the environment \mathfrak{e} will destroy the simple form of Eq. 7, converting an initially pure state to one that is entangled with the environment. Although these interactions are complicated, Sect. 2.1 above suggested that they could be summarized by saying that the subsystem's *phase* is altered by the many environmental particles that interact with it. When we perform the trace operation in Eq. 5, the entanglement leads to the sum of many random phase factors in the off-diagonal elements of ρ , effectively suppressing them within some decoherence time scale T [20, chap. 3]. The diagonal terms are unaffected, however.

We must also extend the simplified discussion above (Eq. 2) by allowing for the possibility that the energies of $|1\rangle$ and $|2\rangle$ may not be exactly equal. Thus, let $H = H_0 + V$, where H_0 is diagonal with eigenvalues E_1 and E_2 and V is the off-diagonal interaction operator appearing in Eq. 2. Eq. 8 then becomes

$$\frac{d\rho_{22}}{dt} = \frac{1}{i\hbar}[V, \rho]_{22} \quad (10)$$

$$\frac{d\rho_{ij}}{dt} = \frac{1}{i\hbar}([V, \rho]_{ij} + (E_i - E_j)\rho_{ij}) - \frac{1}{T}\rho_{ij} \quad \text{for } i \neq j. \quad (11)$$

The environment enters via the last term above, which contains the decoherence time scale T .

The donor can also lose its excitation directly, without transfer of energy to the acceptor. We approximate this effect as a decay term in the equation for ρ_{11} :

$$\frac{d\rho_{11}}{dt} = \frac{1}{i\hbar}[\mathbf{V}, \rho]_{11} - \frac{1}{\tau}\rho_{11}. \quad (12)$$

(Eq. 10 neglects the analogous effect for acceptor deexcitation, which is not relevant to our discussion.) Eqs. 10–12 are sometimes called “Pauli master equations,” or “Redfield equations” [21].

3 FRET

3.1 Formal solution

Förster studied the situation in which the decoherence rate, $1/T$, is much faster than either the transition rate, $\Omega = 2V/\hbar$, or the donor deexcitation rate, $1/\tau$ (the “fast decoherence” limit). Typical numbers for chromophores in solution are $1/T \approx 10^{14} \text{ s}^{-1}$ [22], compared with typical rates for donor fluorescence and mixing $\tau^{-1} \approx \Omega \approx 10^8 \text{ s}^{-1}$. Förster realized that in this situation, the environment effectively supplies a continuum of final states, even though the subsystem states of interest are discrete. This observation may have motivated him to apply the “golden rule,” initially developed for the emission of a photon into an explicit continuum of final states. Rather than appeal to this black box, however, we can equally well proceed simply by expansion in powers of $T\Omega$, as follows.

Let $S = (E_1 - E_2)/\hbar$. Change variables to the four real quantities $U = \rho_{11}$, $W = \rho_{22}$, $X = (\rho_{12} - \rho_{21})/i$, and $Y = \rho_{12} + \rho_{21}$. Then the dynamical equations take the real form

$$\begin{aligned} dU/dt &= -\frac{1}{2}\Omega X - U/\tau \\ dW/dt &= \frac{1}{2}\Omega X \\ dX/dt &= \Omega(U - W) - X/T - SY \\ dY/dt &= -Y/T + SX. \end{aligned} \quad (13)$$

This is a set of coupled linear differential equations with constant coefficients, so its solutions will be combinations of exponentials.

Let $\mathbf{Z}(t)$ be the 4-component vector with entries $U(t)$, $W(t)$, $X(t)$, and $Y(t)$, so that Eq. 13 can be written symbolically as $d\mathbf{Z}/dt = \mathbf{M}\mathbf{Z}$, where \mathbf{M} is a 4×4 matrix. When the coupling $\Omega = 0$, we easily find one solution:

$$\mathbf{Z}_0(t) = \begin{bmatrix} U(t) \\ W(t) \\ X(t) \\ Y(t) \end{bmatrix} = e^{-\beta_0 t} \mathbf{B}_0 \quad \text{where} \quad \mathbf{B}_0 = \begin{bmatrix} 1 \\ 0 \\ 0 \\ 0 \end{bmatrix} \quad \text{and} \quad \beta_0 = 1/\tau. \quad (14)$$

This solution describes spontaneous deexcitation of the donor, for example, via fluorescence.

At small but nonzero Ω , we expand all quantities in powers of $\epsilon = (T\Omega)$. For example, we expand the matrix \mathbf{M} as $\mathbf{M}_0 + \epsilon\mathbf{M}'$, with

$$\mathbf{M}_0 = \begin{bmatrix} -\tau^{-1} & 0 & 0 & 0 \\ 0 & 0 & 0 & 0 \\ 0 & 0 & -T^{-1} & -S \\ 0 & 0 & S & -T^{-1} \end{bmatrix} \text{ and } \mathbf{M}' = \begin{bmatrix} 0 & 0 & -T^{-1}/2 & 0 \\ 0 & 0 & T^{-1}/2 & 0 \\ T^{-1} & -T^{-1} & 0 & 0 \\ 0 & 0 & 0 & 0 \end{bmatrix}.$$

Then we again seek a trial solution to Eqs. 13 that is exponentially changing in time. Expanding the eigenvalue β as $\beta_0 + \epsilon\beta' + \epsilon^2\beta'' + \dots$, the first-order terms in the eigenvalue equation give $(\mathbf{M}_0 + \beta_0\mathbb{I})\mathbf{B}' + (\mathbf{M}' + \beta'\mathbb{I})\mathbf{B}_0 = 0$, while the second-order terms give $(\mathbf{M}_0 + \beta_0\mathbb{I})\mathbf{B}'' + (\mathbf{M}' + \beta'\mathbb{I})\mathbf{B}' + \beta''\mathbf{B}_0 = 0$.

Multiplying both sides of the first-order equation by the transpose \mathbf{B}_0^\dagger and using Eq. 14 gives $\beta' = -\mathbf{B}_0^\dagger\mathbf{M}'\mathbf{B}_0$. Substituting the known \mathbf{B}_0 and \mathbf{M}' gives $\beta' = 0$. At the next order, however,

$$\beta'' = \frac{1}{2} \frac{T^{-1}}{1 + (TS)^2}. \quad (15)$$

Altogether, we find

$$\beta = \tau^{-1} + \frac{\Omega^2 T/2}{1 + (TS)^2} = \tau^{-1} + \frac{2V^2 T}{\hbar^2 + T^2(E_1 - E_2)^2}. \quad (16)$$

3.2 Numerical example

To see the physical implication of Eq. 16, it is instructive to solve Eqs. 13 numerically. If we prepare the initial state $Z(0) = \mathbf{B}_0$, then strictly speaking, the initial rate of decrease of ρ_{11} is the first component of $\mathbf{M}\mathbf{B}_0$, that is, $-1/\tau$. And the initial rate of increase of ρ_{22} is the second component, that is, 0. However, after a brief transient these behaviors change. Figure 1 shows a numerical solution with sample parameter values. We see that, although the time course of ρ_{22} is indeed initially flat, it soon starts to increase exponentially. Similarly, although $\rho_{11}(t)$ initially starts to fall with slope $-1/\tau$, it soon starts to fall as $e^{-\beta t}$, with β given by Eq. 16. These effective first-order rate constants describe the excitation transfer. The second contribution in Eq. 16 shows that the transfer rate has a sharp maximum as a function of the energy difference. Importantly, the area under that peak does not depend on the value of the decoherence time T , as long as T is small enough to justify the approximations made.

3.3 Qualitative discussion

We can now look back and identify the origin of the characteristic physical features of FRET. In general, irreversibility in classical or quantum physics stems from a system “getting lost in phase space” [10]: The probability of returning to an initial state is vanishingly small. For a single atom emitting a photon into vacuum, irreversibility stems from the continuum of available

photon states; in FRET, it stems from entanglement with the surroundings (decoherence).

In the limit of fast decoherence ($T \rightarrow 0$), the third and fourth of Eqns. 13 say that Y rapidly relaxes to zero, whereas the other coherence X adiabatically tracks the quantity $T\Omega(U - W)$. Substituting that value for X into the first two equations shows that, on the resonance $S = 0$, the population difference $U - W$ has a contribution to its time dependence proportional to $T\Omega^2$, as we indeed found (Eq. 16). In words, fast decoherence suppresses the effects of mixed quantum states, but one coherence is constantly “pumped up” by the population difference, and feeds back negatively to it.

3.4 Electric dipole approximation

To apply Eq. 16 to FRET, we now recall that the interaction energy of two electric dipoles is proportional to the product of their electric dipole moments \mathbf{d}_D and \mathbf{d}_A , and to the inverse cube of the distance between them. Thus, in a molecular separation regime where dipole interactions dominate, V in Eq. 16 is proportional to $r^{-3}\langle 2|\mathbf{d}_D \cdot \mathbf{d}_A|1\rangle$, yielding the famous orientation dependence of the FRET rate [23]. The rate is also proportional to r^{-6} , another key feature of FRET [24, 25]. (Note, however, that at very long distances our assumption of instantaneous (electrostatic) interaction fails. More detailed quantum-electrodynamics calculations show a gradual crossover to r^{-2} behavior at large r , as we might have expected naively from the exchange of a real photon [26].)

So far, we have assumed definite (exact) values for the donor’s excited and ground state energies, and similarly for the acceptor. Actually, however, each of these energies changes over time due to molecular motions, that is, changes of the positions of the atomic nuclei. Accordingly, we now introduce realistic (that is, broad) probability distributions of these energies, and average the mean rate for energy transfer over those distributions. The sharply peaked form of Eq. 16 as a function of $E_1 - E_2$ then implies that the mean FRET rate will be proportional to the overlap integral of the two distributions, another key feature of FRET. In fact, in the stated limit Förster was able to find a prediction with *no free parameters* for the FRET rate, in terms of the donor’s measured emission spectrum and fluorescence rate, the acceptor’s measured excitation spectrum and fluorescence cross section, the medium’s index of refraction, and the distance and relative orientation between donor and acceptor.

In particular, the derivation just outlined explains a surprising aspect of FRET, which is that there can be highly specific energy transfer between two particular molecular species, despite the multitude of other directions into which the donor could instead emit a photon, and the crush of other molecules that could instead receive the energy:

- To understand the dominance of FRET over photon emission, note that the “near fields” of a fluctuating dipole fall off with distance as r^{-3} , independent of its frequency. The “radiation fields” fall off more slowly, as r^{-1} , and they do depend on frequency. Turning these statements around, at

small distances the near fields are stronger by a factor of $(\lambda/r)^2$, where λ is the wavelength of light corresponding to donor fluorescence. The square of this ratio can exceed 10^4 .

- Turning to the other nearby molecules, the sharply peaked form of Eq. 16 ensures that only those with a transition resonant with the donor’s emission (overlapping spectra) will have significant probability per unit time to gain energy from it.

3.5 Limitations

Although the approximations made by Oppenheimer and Förster are often excellent for FRET used as a lab technique, later work has shown that they, and the physical picture that they support, do not hold in other situations, such as in photosynthetic apparatus.

First, we assumed that the interaction between two molecules can be approximated as a dipole-dipole interaction. For the tightly spaced photosynthetic chromophores, this approximation is not always valid. Not only does the multipole approximation break down, but “exchange interactions” (direct contact between the fluorophores’ electron clouds) start to be significant—the Dexter mechanism.

Second, the derivation of Eq. 16 assumed that excitation transfer is much slower than quantum decoherence. In that situation, we got a simple rate law for excitation transfer, and a nice picture of localized excitations. But there is a hierarchy of substructures in the photosynthetic apparatus, and within some of them the transfers are extremely fast. In this situation, it makes more sense to regard a whole array of chromophores as a single “supermolecule,” with delocalized excitations called “Frenkel excitons.” The supramolecular units in turn transfer excitons among themselves via FRET-like processes. For more about these systems, see [27–33].

4

FRET will continue to be the technique of choice for many more advances in single-molecule and single-cell biophysics, including some not yet imagined. This Perspective has pointed out that viewed as a physical phenomenon, FRET displays aspects not familiar from undergraduate (or even graduate) training in physics or chemistry, and these are key to its usefulness. Indeed, there were many missteps along the way to understanding FRET [9, 12, 13], but I have argued that an appreciation of the role of quantum decoherence leads directly to the main features.

Acknowledgments

This work was partially supported by the United States National Science Foundation under Grant PHY-1601894. Some of the work was done at the Aspen Center for Physics, which is supported by NSF grant PHY-1607611.

References

- [1] Hwang, L. C., J. Hohlbein, S. J. Holden, and A. N. Kapanidis. 2009. Single-molecule FRET: Methods and biological applications. In *Handbook of Single-Molecule Biophysics*. P. Hinterdorfer and A. van Oijen, editors. Springer, New York, pp. 129–164.
- [2] Ha, T., T. Enderle, D. F. Ogletree, D. S. Chemla, P. R. Selvin, and S. Weiss. 1996. Probing the interaction between two single molecules: fluorescence resonance energy transfer between a single donor and a single acceptor. *Proc. Natl. Acad. Sci. USA*, 93:6264–6268.
- [3] Chang, A. C., A. H. Mekhdjian, M. Morimatsu, A. K. Denisin, B. L. Pruitt, and A. R. Dunn. 2016. Single molecule force measurements in living cells reveal a minimally tensioned integrin state. *ACS Nano*, 10:10745–10752.
- [4] Oppenheimer, J. R. 1941. Internal conversion in photosynthesis. *Phys. Rev.*, 60:158.
- [5] Förster, T. 1948. Zwischenmolekulare energiewanderung und fluoreszenz. *Ann. Physik*, 6:55–75. English translation in [7].
- [6] Arnold, W. and J. R. Oppenheimer. 1950. Internal conversion in the photosynthetic mechanism of blue-green algae. *J. Gen. Physiol.*, 33:423–435.
- [7] Förster, T. 1993. Intermolecular energy migration and fluorescence. In *Biological Physics*. E. V. Mielczarek, E. Greenbaum, and R. S. Knox, editors. AIP Press, New York, pp. 148–160.
- [8] Dirac, P. A. M. 1927. The quantum theory of the emission and absorption of radiation. *Proc. Roy. Soc. Lond. A*, 114:243–265.
- [9] Clegg, R. M. 2006. The history of FRET. In *Reviews in Fluorescence 2006*, volume 2006 of *Reviews in Fluorescence*. C. Geddes and J. Lakowicz, editors. Springer US, New York, pp. 1–45.
- [10] Nitzan, A. 2006. *Chemical Dynamics in Condensed Phases: Relaxation, Transfer and Reactions in Condensed Molecular Systems*. Oxford Univ. Press, Oxford UK.
- [11] Agranovich, V. M. and M. D. Galanin. 1982. *Electronic Excitation Energy Transfer in Condensed Matter*. Elsevier North-Holland, New York.

- [12] Masters, B. R. 2014. Paths to Förster’s resonance energy transfer (FRET) theory. *Eur. Phys. J. H*, 39:87–139.
- [13] Knox, R. S. 2012. Förster’s resonance excitation transfer theory: not just a formula. *J. Biomed. Opt.*, 17:011003.
- [14] Nelson, P. 2017. *From Photon to Neuron: Light, Imaging, Vision*. Princeton Univ. Press, Princeton NJ.
- [15] Feynman, R. P., R. Leighton, and M. Sands. 2010. *The Feynman Lectures on Physics*, new millennium edition, volume 3. Basic Books, New York. Free online: <http://www.feynmanlectures.caltech.edu/>.
- [16] Jang, S. 2007. Generalization of the Förster resonance energy transfer theory for quantum mechanical modulation of the donor-acceptor coupling. *J. Chem. Phys.*, 127:174710.
- [17] Jang, S., H. Hossein-Nejad, and G. D. Scholes. 2014. Generalized Förster resonance energy transfer. In *Quantum Effects in Biology*. M. Mohseni, Y. Omar, G. S. Engel, and M. B. Plenio, editors. Cambridge Univ. Press, Cambridge UK, pp. 53–81.
- [18] Schumacher, B. and M. D. Westmoreland. 2010. *Quantum Processes, Systems, and Information*. Cambridge Univ. Press, Cambridge UK.
- [19] Berman, P. R. and V. S. Malinovsky. 2011. *Principles of Laser Spectroscopy and Quantum Optics*. Princeton Univ. Press, Princeton NJ.
- [20] Schlosshauer, M. A. 2007. *Decoherence and the Quantum-To-Classical Transition*. Springer, New York.
- [21] Silbey, R. J. 2011. Description of quantum effects in the condensed phase. *Procedia Chem.*, 3:188–197.
- [22] Gilmore, J. and R. H. McKenzie. 2008. Quantum dynamics of electronic excitations in biomolecular chromophores: Role of the protein environment and solvent. *J. Phys. Chem. A*, 112:2162–2176.
- [23] Iqbal, A., S. Arslan, B. Okumus, T. J. Wilson, G. Giraud, D. G. Norman, T. Ha, and D. M. J. Lilley. 2008. Orientation dependence in fluorescent energy transfer between Cy3 and Cy5 terminally attached to double-stranded nucleic acids. *Proc. Natl. Acad. Sci. USA*, 105:11176–11181.
- [24] Sindbert, S., S. Kalinin, H. Nguyen, A. Kienzler, L. Clima, W. Bannwarth, B. Appel, S. Müller, and C. A. M. Seidel. 2011. Accurate distance determination of nucleic acids via Förster resonance energy transfer: implications of dye linker length and rigidity. *J. Am. Chem. Soc.*, 133:2463–2480.
- [25] Wozniak, A. K., G. F. Schröder, H. Grubmüller, C. A. M. Seidel, and F. Oesterhelt. 2008. Single-molecule FRET measures bends and kinks in DNA. *Proc. Natl. Acad. Sci. USA*, 105:18337–18342.

- [26] Salam, A. 2010. *Molecular quantum electrodynamics: Long-range intermolecular interactions*. Wiley, Hoboken NJ.
- [27] Jang, S. and Y. Cheng. 2013. Resonance energy flow dynamics of coherently delocalized excitons in biological and macromolecular systems: Recent theoretical advances and open issues. *WIREs Comput. Mol. Sci.*, 3:84–104.
- [28] Strümpfer, J., M. Şener, and K. Schulten. 2012. How quantum coherence assists photosynthetic light harvesting. *J. Phys. Chem. Lett.*, 3:536–542.
- [29] Şener, M., J. Strümpfer, J. Hsin, D. Chandler, S. Scheuring, C. N. Hunter, and K. Schulten. 2011. Förster energy transfer theory as reflected in the structures of photosynthetic light-harvesting systems. *ChemPhysChem*, 12:518–531.
- [30] Engel, G. S. 2011. Quantum coherence in photosynthesis. *Procedia Chem.*, 3:222–231.
- [31] Beljonne, D., C. Curutchet, G. D. Scholes, and R. J. Silbey. 2009. Beyond Förster resonance energy transfer in biological and nanoscale systems. *J. Phys. Chem. B*, 113:6583–6599.
- [32] Ishizaki, A. and G. R. Fleming. 2009. Unified treatment of quantum coherent and incoherent hopping dynamics in electronic energy transfer: Reduced hierarchy equation approach. *J. Chem. Phys.*, 130:234111.
- [33] Seibt, J. and T. Mancal. 2017. Ultrafast energy transfer with competing channels: Non-equilibrium Förster and Modified Redfield theories. *J. Chem. Phys.*, 146:174109.

Caption

Figure 1: Diagonal elements of the density matrix (“populations”) with sample parameter values $\Omega T = 0.05$, $T/\tau = 0.01$, and $ST = 0.05$. These values are less extreme than the realistic ones mentioned in Sect. 3.1; they were chosen to make the crossover behavior visible in the graph. For comparison, the dotted line shows a single exponential that asymptotically matches the acceptor population at long times.